

## LC-MS Compatible Stability Indicating RP-UPLC Method for the Estimation of Ester Prodrug of Mycophenolic Acid in Injection Formulation

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This study explores the ultra performance liquid chromatography for the quantification of mycophenolate mofetil (an immunosuppressant agent) in an injection formulation. Mycophenolate mofetil was separated from its seven specified impurities on HSS (high strength silica) C18 column (100 × 2.1 mm, 1.8 μm). The optimized mobile phase is compatible to mass spectrometry and hence the method can be equally run on LC-MS system for mass determinations. Efficient UV detection at 250 nm enabled determination of mycophenolate mofetil with no interference from placebo solution, diluent and other specified and unspecified impurities. The retention time of mycophenolate mofetil in the method was 1.79 min and all other impurities were eluted within 8 min. The method optimization studies show that the selectivity of mycophenolate mofetil and its impurities was dependent on the type of organic solvent and its elution strength. The method was validated for specificity, linearity, precision, accuracy and robustness in range of 50 to 150 % of target analyte concentration (i.e. 62.50-195.31 μg mL<sup>-1</sup>). The linearity of peak area responses *versus* concentrations was demonstrated by linear least square regression analysis. An excellent linear relation exists between the concentration and detector response (R<sup>2</sup> > 0.9999). Precision of the test method was proved with a relative standard deviation less than 1.0 %.

**Keywords:** UPLC, Mycophenolate mofetil, Stability, Indicating assay, Validation.

### INTRODUCTION

Mycophenolate mofetil is an immunosuppressant drug [1]. Immunosuppressant drugs belong to a class of drugs that reduce or suppress the immune strength of the body. These are used during organ transplantation or for an autoimmune disorder [2]. Mycophenolate mofetil is an ester prodrug of mycophenolic acid (MPA), which is the main active moiety. Other drugs, which are employed in immunosuppressive therapy are cyclosporine A, azathioprine, prednisone FK-506 (Tacrolimus), Rapamycin (Sirolimus), *etc.* Though cyclosporine A was a choice of drug among the immunosuppressants in 1970's, it is associated with certain side effects like hypertension, hypercholesterolemia, nephrotoxicity, diabetogenic effects and neurotoxicity. Azathioprine and prednisone were found ineffective in providing adequate immunosuppression for patients having complex kidney problems and who further require liver or heart transplantation. Among these immunosuppressants mycophenolate mofetil has emerged as a novel potent drug with enhanced bioavailability and reduced toxicities that are associated with the above mentioned drugs [3,4].

Mycophenolate mofetil is a white to off white crystalline substance with an empirical formula C<sub>23</sub>H<sub>31</sub>NO<sub>7</sub>. Its molecular weight is 433.49. Its chemical name is (*E*)-2-morpholinoethyl (*E*)-6-(4-hydroxy-6-methoxy-7-methyl-3-oxo-5-phthalanyl)-4-methyl-4-hexenoate. It is slightly soluble in water. The reported solubility in water is 43 mg mL<sup>-1</sup> [5,6]. It has two pKa values, 5.6 and 8.5 for morpholine group and phenolic group respectively. The existing dosage forms of mycophenolate mofetil are tablets, capsules, oral suspension and injection [7]. The current study utilizes its injection formulation for new analytical method.

Mycophenolate mofetil is a pro-drug and a morpholino ethyl ester of mycophenolic acid (MPA). In 1913, MPA was first isolated from the culture of *Penicillium stoloniferum* [8]. It was then reported as a metabolic product from *Penicillium brevicompactum*, *Penicillium brevicompactum* and *Penicillium stoloniferum*, which also belong to same series of *Penicillium* [9].

The chemical structure of MPA was determined in 1952, while its crystal structure was reported in 1972 [10,11].

Mycophenolate is official in United States Pharmacopoeia (USP) and European Pharmacopoeia (Ph. Eur) and Indian Pharmacopoeia [12-17]. The potential impurities of myco-

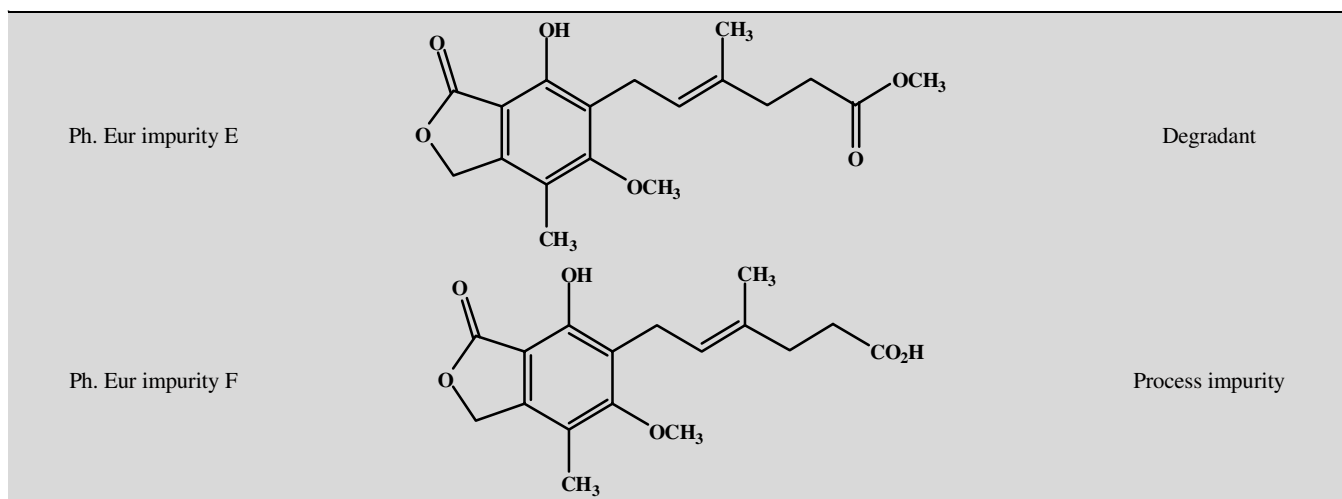
phenolate mofetil are listed in Table-1 [16]. USP compendia contains monographs of different dosage forms (mycophenolate mofetil for injection, capsules, tablets, delayed release tablets and oral suspension).

Ph. Eur monographs contain titrimetric method and HPLC method for assay of mycophenolate mofetil and related substances, respectively. The mobile phase consists of a mixture of acetonitrile and water (35:65 v/v) containing 2 mL of triethylamine (pH 5.3 with dil.  $H_3PO_4$ ). The separation of sample components was carried out on C8 (octylsilyl) column with

dimensions 250 mm  $\times$  4.6 mm, 5  $\mu$ m. The flow rate was 1.5 mL  $min^{-1}$  and detection wavelength was 250 nm. USP monograph for mycophenolate mofetil prescribes a common method for assay of it and its organic impurities. The method follows an isocratic separation with a mixture of acetonitrile and triethylamine buffer (pH 5.3) in the ratio 7:13 v/v. The column and other chromatographic conditions are like the above stated method in Ph. Eur monograph. A separate method is available for assay of it and its organic impurities present in USP for injection formulation. The method comprises of a triethyl-

TABLE-1  
NAMES, CHEMICAL STRUCTURES AND NATURE OF MYCOPHENOLATE IMPURITIES [Ref. 16]

Name	Chemical structure	Nature
Ph. Eur Impurity A USP related compound A		Process impurity
USP related compound B		Degradant
Ph. Eur impurity B		Process impurity
Ph. Eur impurity C		Process impurity
Ph. Eur impurity D		Process impurity



amine buffer (pH 7.2) mixed with acetonitrile in the ratio 7:3 v/v, as the mobile phase. The separation was carried out on phenyl (L11) column (250 mm × 4.6 mm, 5 μm). A similar method consisting of triethylamine buffer and L11 column is available in USP for the determination of mycophenolate mofetil in oral suspension. Similarly, methods are available in USP for the determination of mycophenolate mofetil in tablets and capsule formulations. These methods also use the same mobile phase containing triethylamine and acetonitrile, C8 column as described in, for other formulations in USP. Indian Pharmacopoeia publishes two methods for mycophenolate mofetil API and capsules dosage form IP. First method is similar to that of Ph. Eur monograph. C<sub>18</sub> column of dimensions 250 mm × 4.6 mm, 5 μm was used for the analysis of related substances. The assay of mycophenolate mofetil in IP monograph is estimated using a titrimetric method.

Thorough literature survey revealed few reported methods on the determination of mycophenolate mofetil by HPLC and UPLC methods. One HPLC method was reported on the determination of mycophenolate mofetil in human plasma [18]. The method uses a mobile phase containing a buffer (pH 3.6) and acetonitrile mixed in the ratio 59:41 v/v. The buffer contains citratephosphate (0.05 M) and heptanesulphonic acid (0.02 M). The mobile phase was pumped at 0.4 mL min<sup>-1</sup> and detection was carried at 254 nm. The separation was performed on a C<sub>18</sub> column (150 mm × 4.6 mm, 5 μm). Mycophenolate mofetil elutes at about 15 min and the total run time of the method was 25 min. An isocratic HPLC method is available for the determination of mycophenolate mofetil in pure and tablets dosage form [19]. The method comprises of phosphate buffer (0.03 M containing 1 mL of triethylamine, pH 2.9) mixed with acetonitrile in the ratio 60:40 v/v as the mobile phase. The separation was carried out on a C<sub>18</sub> column with dimensions 250 mm × 4.6 mm, 5 μm. The method lacks discussion on specificity of listed impurities and forced degradation studies.

One reversed phase UPLC method was reported on quantitation of mycophenolate mofetil in tablets dosage form [20]. The separation was carried on a symmetry C<sub>18</sub> column (100 mm × 4.6 mm, 3.5 μm) using a mobile phase composed of phosphate buffer and acetonitrile mixed in the ratio 35:65 v/v. Mycophenolate mofetil in tablets was detected at 228 nm. The method lacks discussion on forced degradation studies and no

specified impurities were discussed in specificity. Another UPLC method was reported on forced degradation study and quantification of mycophenolate in tablets [21]. This method was developed on BEH column having dimensions 2.1 mm × 100 mm, 1.7 μm. The mobile phase used in the method was a combination of phosphate buffer and acetonitrile (35:65 v/v) pumped at a flow rate of 0.4 mL min<sup>-1</sup>. Few HPLC methods were found on the analysis of mycophenolate acid and mycophenolate acid glucuronide in biosamples [22-31]. These methods used similar mobile phases as reported above and carried out separations on conventional reversed phase columns coupled with mass spectrophotometer. The literature reveals that no single short method is reported on the estimation of mycophenolate mofetil in injection formulation in presence of its potential impurities. Hence, an attempt is made to develop a shorter method which is also mass compatible.

## EXPERIMENTAL

The HPLC grade acetonitrile and A.R. grade ammonium acetate used in the preparation of mobile phase were obtained from Merck, India. The reference standards and impurity standards were procured from USP and Ph.Eur.

An integrated Acquity UPLC system is from Waters Corporation, Milford, USA and equipped with a Waters photodiode array detector (PDA). Data collection and analysis were performed using the Empower software 2pro (Waters Corporation). Balances used for weighing the reference standards and samples were from Mettler Toledo. The columns HSS T3 was procured from Waters India Pvt. Ltd. Class A glasswares used in conducting the experiments and validations was from Duran and Borosil, India.

### Standard and sample preparations

**Diluent preparation:** A mixture of acetonitrile and water (90:10 v/v) was used for the preparation of diluent, standards, system suitability solutions and sample solution. This selection was based on the solubility studies conducted on the active drug substance.

**Preparation of mycophenolate working standard solution:** Accurately weighed and transferred 39 mg of mycophenolate working standard into 50 mL volumetric flask. Added 25 mL of the diluent and sonicated for 2 min to dissolve the

contents. Made up to the mark with the diluent. Further transfer 4 mL of the above solution into 25 mL volumetric flask and made up to the mark with the diluent to attain a concentration of 125  $\mu\text{g mL}^{-1}$ .

**Preparation of mycophenolate impurity mixed stock solution:** Accurately weighed and transferred 2.5 mg of each impurity of mycophenolate into 100 mL volumetric flask. Added 25 mL of the diluent and sonicated for 2 min to dissolve the contents. Made up to the mark with the diluent.

**Preparation of sample solution (mycophenolate for injection 6 mg/mL):** Taken five vials of mycophenolate powder for injection, reconstituted each vial as recommended in the PIL (Product Information Leaflet). Pooled the contents of all three vials. Pipetted 5 mL of the sample into 25 mL volumetric flask. Added 15 mL of the diluent and sonicated for 2 min to mix the contents. Made up to the mark with the diluent. Further diluted 2.6 mL of the above solution to 25 mL with diluent and mixed well to get the final concentration of 125  $\mu\text{g mL}^{-1}$ .

**Preparation of sample solution spiked with impurity mixture:** Pipetted 5 mL of reconstituted sample into 25 mL volumetric flask. Added 15 mL of diluent and sonicated for 2 min to mix the contents. Made up to the mark with the diluent. Further pipetted 2.6 mL of the above solution and 2.5 mL of the impurity mixed stock solution into 25 mL volumetric flask and diluted up to the mark with the diluent. Mixed well to get a final concentration of 2 % of impurities.

**Preparation of placebo solution:** Transferred 5 mL of the placebo solution to 25 mL volumetric flask. Added 15 mL of the diluent and sonicated for 2 min to mix the contents. Made up to the mark with the diluent. Further diluted 2.6 mL of the above solution to 25 mL with diluent and mixed well.

## RESULTS AND DISCUSSION

**Method development and optimization:** The diluent used for preparing the standard, system suitability, impurity standards sample and placebo solution was based on the solubility studies conducted for mycophenolate as per USP general notices (5.30. Description and solubility) (USP 40-NF35) [32]. Based on the obtained data and sample matrix, a mixture of acetonitrile and water (90:10 v/v) was selected as the diluent

The detection wavelength was determined by injecting a detectable concentration of USP mycophenolate mofetil reference standard into the chromatographic system having photodiode array (PDA) detector. The obtained spectrum exhibits maxima at 304.3, 250.1 and at 215.3 nm (Fig. 1). However, owing to the possible interferences from organic solvents and inactive excipients the wavelength 215 nm was not considered. 304.3 nm has lesser absorbance than 250.1 nm. Hence based on the UV spectra, literature and USP, Ph. Eur monographs and required sensitivity, 250 nm was selected as the  $\lambda_{\text{max}}$  for detection of mycophenolate mofetil.

**Method development (selection of chromatographic conditions):** The method was developed using the principles of reversed phase chromatography (RPC). The basis for choosing the RP chromatography is the nature and type of the selected compound. Mycophenolate mofetil is a low molecular weight, ester pro-drug of mycophenolic acid (MPA) [8] and hence is a neutral molecule. Reversed phase chromatography is the first

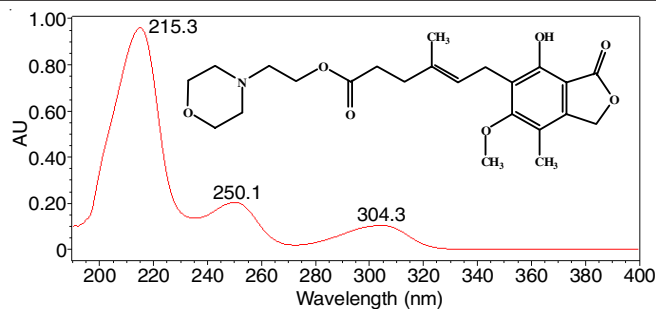


Fig. 1. Ultraviolet-visible spectrum of mycophenolate

choice for neutral molecules having molecular weight less than 2000 Da [33,34].

**Trial-1:** For initial development trials, the most common RP column *i.e.*, C18 column with dimensions 150 mm  $\times$  4.6 mm and particle size 5  $\mu\text{m}$  was chosen. C18 (octadecyl silyl) is the most common RP stationary phase in which long chain alkyl groups are bonded to silica. These are rugged and operate in a wide pH range (pH 2-8). Since mycophenolate mofetil is hydrophobic in nature, pure water cannot be used as an eluent. This is because water being highly polar, repels the hydrophobic molecules and hence the retention times are delayed [35]. The first development strategy contains a mixture of aqueous and non-aqueous mobile phase *i.e.*, Buffer: acetonitrile (80:20 v/v). Ammonium acetate was chosen as the buffer (0.01 M) at pH 5.8 (based on mycophenolate mofetil pKa 6.19) [5]. Acetonitrile was chosen based on the molecule solubility for diluent compatibility. The flow rate of the pump was to set 1.0 mL  $\text{min}^{-1}$ . The initial column oven temperature was set at 30  $^{\circ}\text{C}$ . With these set of chromatographic conditions, a mixture of mycophenolate mofetil reference standard spiked with USP impurities and Ph. Eur. Peak identification mixture (20  $\mu\text{L}$ ) was injected into the chromatograph. The chromatogram obtained from trial 1 is shown in Fig. 2. Three impurities were separated from the main compound and remaining were found merged with mycophenolate mofetil peak.

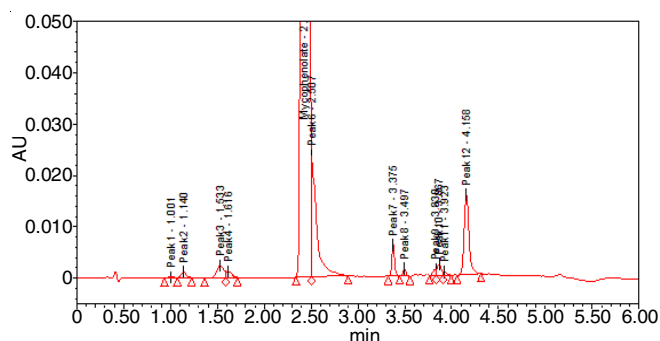


Fig. 2. Chromatogram obtained from Trial-1

**Trial-2:** Since the structures of few impurities are similar to one another, isocratic mode of separation was ineffective in separating all impurities from the principal peak (mycophenolate mofetil). Hence in the next trial, a gradient programme-1 (T (min)/%B: 0/20, 0.6/25, 0.8/30, 0.9/25, 1.2/20, 5/20) was introduced to improve separation from the principal peak and the mobile pH was modified by 0.5 units (pH 4.5). The remaining chromatographic conditions were retained as such. The obtained chromatogram with trial-2 is shown in Fig. 3.



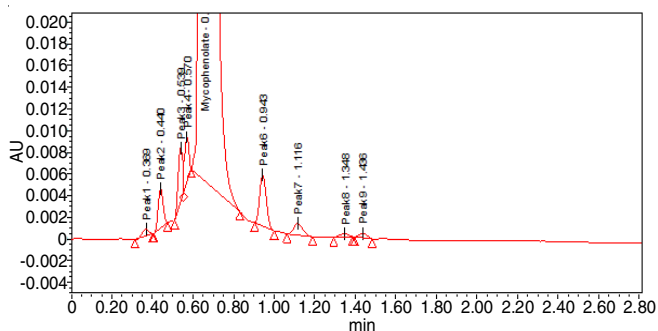


Fig. 3. Development chromatogram obtained from Trial-2

The inference was that three more impurities were resolved from mycophenolate mofetil peak with resolution  $< 1.0$ . From the experiment, it can be observed that an increase in of 10 % acetonitrile from 0 min to 0.8 min, has resulted in improved resolution between the analytes and their selectivity ( $\alpha$ ). This indicates that the analyte peaks are sensitive to organic modifier in the mobile phase.

**Trial-3:** An attempt was made by changing the organic modifier from acetonitrile to methanol. The remaining chromatographic conditions including the gradient programme-1 was retained as such. The obtained chromatogram from trial-3 is shown in Fig. 4. Again, few peaks were observed merging with the principal peak, resolution was reduced between USP related compound B and principal peak. The selectivity differs with respect to the acidity, basicity and polarity index of methanol (PI = 5.1) and acetonitrile (PI = 5.8).

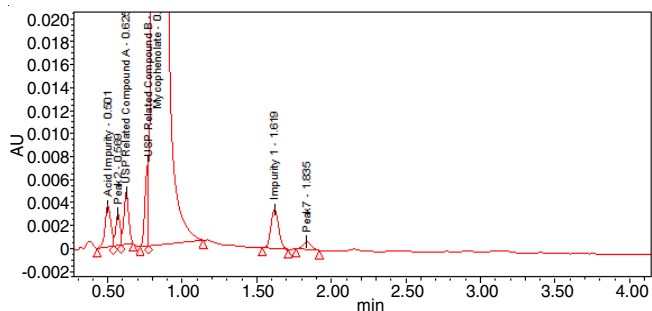


Fig. 4. Development chromatogram obtained from Trial-3

**Trial-4 and method optimization studies:** In trial-4 acetonitrile was retained in place of methanol and the column C18 was replaced with another brand of C18 (HSS C18) with higher carbon (15 %) content [36,37]. Literature shows that a higher carbon loading and smaller particle size effect the selectivity and which in turn effects the resolution between the sample components. Hence, the selectivity increases with an increase in carbon content [38]. The new column dimensions are 100 mm  $\times$  2.1 mm, 1.8  $\mu$ m. Here the pore size is reduced from 5 to 1.8  $\mu$ m. Different flow rates were studied to understand the chromatographic pattern (0.3, 0.5, 0.8 and 1.0 mL min<sup>-1</sup>). The effect of temperature on the separation was studied at 35, 40, 45 and 50 °C. Separation was achieved for all the peaks and also between the USP related compound B and mycophenolate was  $> 2.0$ . Optimal resolution was achieved at column oven temperature 45 °C and flow rate 0.3 mL min<sup>-1</sup>. Impurity 2 which was late eluting in previous trials has eluted

at higher organic composition (a gradient programme-3, (T (min)/%B: 0/60, 3/60, 4/80, 6/80, 6.5/60, 9/60). The optimized chromatogram is shown in Fig. 5, while an un-zoomed chromatogram is shown in Fig. 6. The optimized chromatographic conditions are summarized as: Flow rate: 0.3 mL min<sup>-1</sup>, wavelength: 250 nm, injection volume: 10  $\mu$ L; column oven temperature: 45 °C; mobile phase A: 0.02 M ammonium acetate (pH 4.5), mobile phase B: acetonitrile.

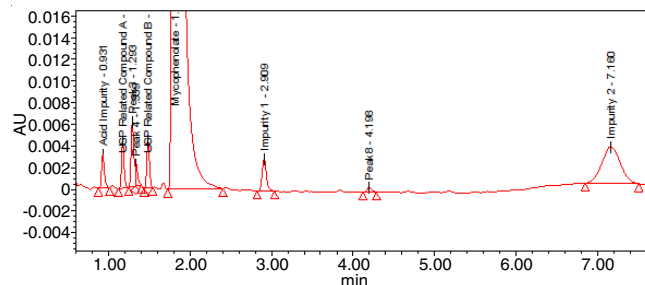


Fig. 5. Development chromatogram obtained from Trial-4

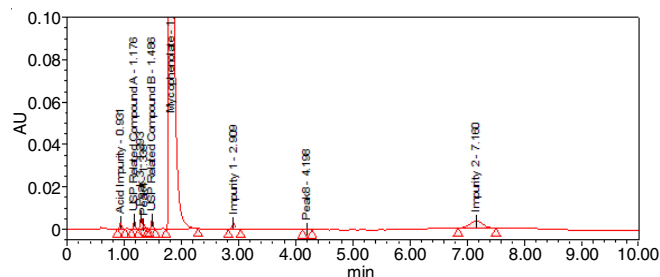


Fig. 6. Development chromatogram obtained from optimized chromatographic conditions

**Method validation:** The developed and optimized UPLC method was taken up for validation. The method validation was carried out in accordance with the validation guidelines (Q2R) recommended by ICH. The developed UPLC method was validated with respect to system suitability, specificity, accuracy, precision, linearity, limit of detection, limit of quantitation and robustness. Each validation parameter is explained in detailed in below sections.

**System suitability and system precision:** The system suitability and system precision for the intended analysis were evaluated from five and six replicate injections respectively of standard solution. The typical system suitability chromatogram is shown in Fig. 7. The system suitability parameters like USP tailing factor, USP plate counts and % RSD for mycophenolate mofetil peak were determined and evaluated (Table-2).

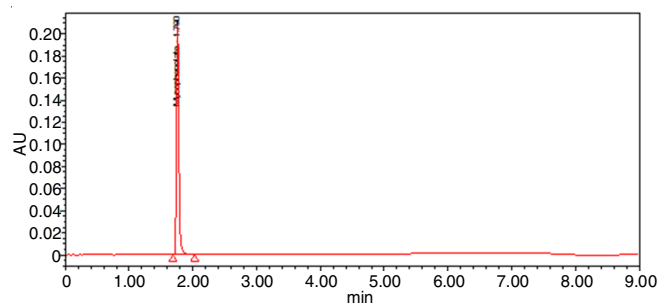


Fig. 7. Typical chromatogram of system suitability

TABLE-2  
RESULTS OF SYSTEM SUITABILITY AND SYSTEM PRECISION

Peak name	Retention time (min)	USP plate counts	USP tailing	%RSD*	System precision %RSD**
Mycophenolate	1.790	23710	1.0	0.2	0.3

\*Data from five replicate injections of standard solution; \*\*Data from six replicate injections of standard solution

### Specificity

**Diluent interference:** The diluent was injected as such into the chromatograph to assess its interference (if any) at the retention time of mycophenolate mofetil. Mycophenolate mofetil elutes at 1.790 min and there is no interference at the retention time of 1.790 min due to the diluent (Fig. 8).

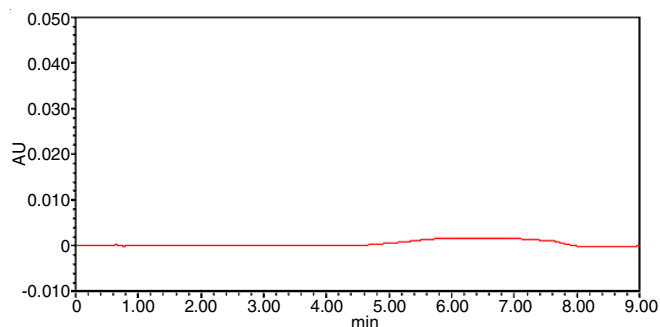


Fig. 8. Typical chromatogram of diluent

**Placebo interference:** Placebo solution was prepared as per the optimized methodology and injected into the chromatographic system. Mycophenolate mofetil elutes at 1.790 min and there is no interference at the retention time of 1.790 min due to the placebo (Fig. 9).

**Interference due to specified impurities of mycophenolate:** To assess the interference due the specified impurities of

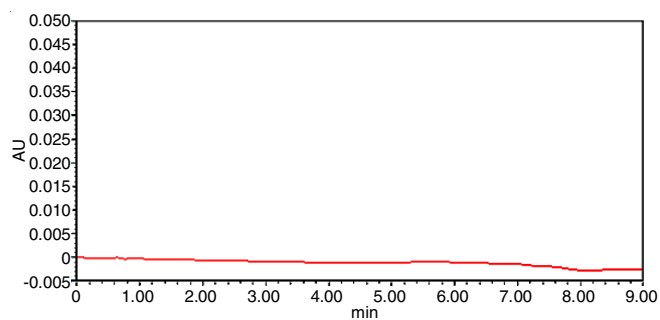


Fig. 9. Typical chromatogram of placebo

mycophenolate, a sample solution was spiked with small portion (2%) of listed impurities (Table-1) by injecting into the chromatographic system. The mycophenolate mofetil peak was evaluated for the homogeneity by measuring the purity angle and purity threshold using the empower software (Figs. 10-13). Assay of mycophenolate mofetil in presence of specified impurity mixture was calculated and compared with that of unspiked sample (Table-3). Assay of the sample spiked with impurities was found 99% and that of pure sample was 98.8%. The absolute difference (0.2) is very small and indicates that the assay of mycophenolate was unaffected in presence of its impurities. The purity data indicate that the purity angle is less than the purity threshold for all three samples, which concludes that mycophenolate mofetil peak is homogenous and the new method is analyte specific.

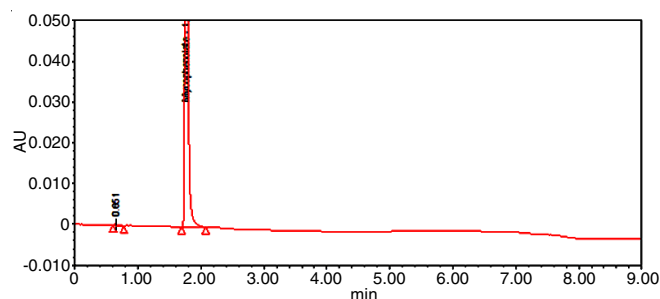


Fig. 10. Typical chromatogram of unspiked sample

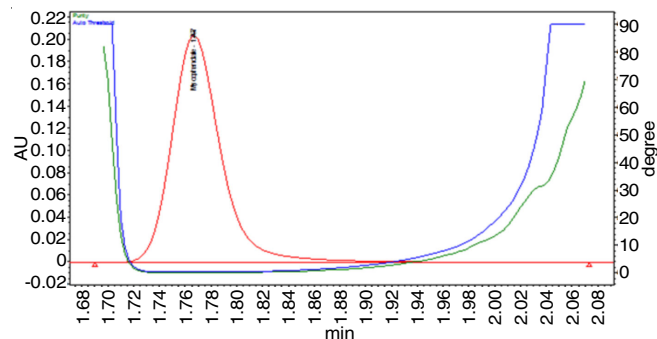


Fig. 11. Purity plot for mycophenolate peak (from unspiked sample)

TABLE-3  
COMPARISON OF ASSAY RESULTS AND PEAK PURITY RESULTS OF SAMPLE SPIKED WITH IMPURITY AND UNSPIKED SAMPLE

Sample Details	Assay (%)	Purity angle	Purity threshold	Purity flag*
Unspiked sample-1	99.4	0.127	0.345	No
Unspiked sample-2	97.9	0.332	0.484	No
Unspiked sample-3	99.2	0.135	0.242	No
% Average	98.8	NA	NA	No
Spiked sample-1	99.1	0.243	0.274	No
Spiked sample-2	99.0	0.236	0.347	No
Spiked sample-3	98.9	0.136	0.346	No
%Average assay	99.0	NA	NA	NA

Difference between average assays of unspiked and spiked sample = 0.2

\*Purity flag 'No' indicates, peak is homogenous (Empower software); Peak is homogeneous if purity angle is less than purity threshold

TABLE-4  
RESULTS OF STRESS STUDY AND PEAK PURITY DATA

Parameter	Stress Conditions	%Assay of degraded sample A	Degradation w.r.t. control B*	Purity angle	Purity threshold
Control sample	No exposure	99.1	NA	0.143	0.302
Acid hydrolysis	1 mL of 1N HCl for 3 h at room temperature	95.4	3.7	0.046	0.325
Base hydrolysis	1 mL of 0.5N NaOH for 3 h at room temperature	31.6	68.1	0.135	0.369
Oxidation	1 mL of 30 % H <sub>2</sub> O <sub>2</sub> for 3h at room temperature	89.2	10.0	0.048	0.301
Thermal degradation	60 °C for 5 h	97.1	2.0	0.047	0.304
Photolytic degradation (UV)	200 Watt h/m <sup>2</sup>	98.7	0.4	0.127	0.311
Photolytic degradation (light)	1.2 Million lux hours	99.1	0.0	0.114	0.320

$$*B = (99.1 - A)/99.1 \times 100$$

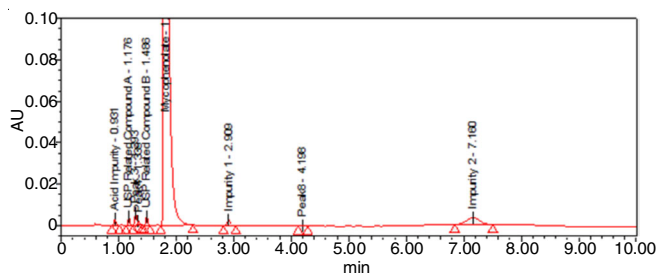


Fig. 12. Typical chromatogram of sample spiked with impurities

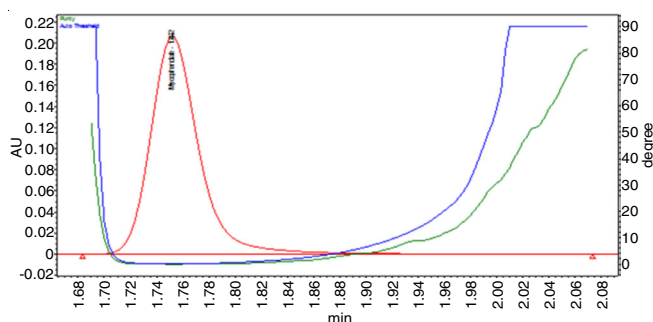


Fig. 13. Purity plot for mycophenolate (from impurity spiked sample)

**Interference from degradation products:** To assess the interference due to the degradation products, mycophenolate mofetil injection sample and placebo solutions were exposed to various stress conditions. The stressed and neutralized samples were then diluted with the diluent to obtain a concentration of about 125  $\mu\text{g mL}^{-1}$ . The obtained solutions were chromatographed as per the optimized methodology. All chromatograms were processed by using the Empower 2pro software. The homogeneity of mycophenolate peak from the stressed samples was evaluated from the purity angle and peak threshold data using the PDA detector. The stress parameters, stress conditions, % degradation in each stress parameter are listed in Table-4. The stress study data indicate that highest degradation was observed in base hydrolysis (68.1 %), followed by oxidation (10 %), acid hydrolysis (3.7 %) and thermal degradations (2 %). The purity angle was less than the purity threshold in all the stress conditions which indicates that mycophenolate peak is homogeneous in presence of its degradation impurities and the method is stability indicating.

**Linearity:** Linearity of detector response against the standard concentration was demonstrated from 25 to 150 % of 125  $\mu\text{g mL}^{-1}$  using minimum six calibration levels (25, 50, 75, 100, 125 and 150 %). The linearity solutions were prepared from a

standard stock solution by appropriate dilutions and each solution was chromatographed. The area response was recorded and the data was evaluated by using linear regression method. Tables 5 and 6 summarizes the linearity results and regression parameters. The obtained data shows an excellent correlation ( $R^2 > 0.9999$ ) between the peak area and mycophenolate concentration which indicates that the method follows a linear fit model.

TABLE-5  
RESULTS OF LINEARITY

S. No.	Level (%)	Concentration ( $\mu\text{g mL}^{-1}$ )	Average area counts ( $\mu\text{v}^*\text{sec}$ )
1	25	31.25	133036
2	50	62.5	263021
3	75	93.75	399462
4	100	125.0	532632
5	125	156.25	666808
6	150	195.3141	830176

TABLE-6  
SUMMARY OF REGRESSION PARAMETERS

S. No.	Parameter	Obtained values
1	Correlation coefficient	0.9999
2	Slope	4263.2
3	Y-Intercept	-987.51
4	Residual sum of squares	12754809

**Accuracy:** The study of accuracy was performed by recovery method using a known amount of active pharmaceutical ingredient (mycophenolate mofetil API) in fixed amount of placebo. Samples were prepared as per the proposed method at three levels *i.e.*, 50, 100 and 150 % of target analyte concentration (125  $\mu\text{g mL}^{-1}$ ) in triplicate ( $n = 3$ ). Each preparation was injected in duplicate ( $n = 2$ ) into the chromatographic system. For each recovery level, % mean recovery was calculated (Table-7). Mean and % relative standard deviation of all three recovery levels were calculated and assessed for accuracy of the method. The % mean recovery was between 98.6 and 100.0, which is well within the acceptance criteria. The relative standard deviations of all three levels are between 0.2 to 0.6 %, which indicates that the method can extract analyte from the sample matrix accurately and precisely.

**Method precision and intermediate precision (ruggedness):** The repeatability of the method was established by preparing and injecting six samples as per the optimized methodology (125  $\mu\text{g mL}^{-1}$ ). Intermediate precision was carried out by analyzing six samples on different instrument, different

TABLE-7  
RESULTS OF ACCURACY

S. No.	Recovery levels	Amount added (mg)	Amount recovered (mg)	Recovered (%)	Mean recovery (%)	RSD (%)
1	50 %	19.59	19.29	98.5	98.6	0.2
2	50 %	19.59	19.36	98.8		
3	50 %	19.59	19.28	98.4		
1	100 %	39.06	39.38	100.6	100.0	0.6
2	100 %	39.06	38.97	99.8		
3	100 %	39.06	38.87	99.5		
1	150 %	58.5	57.95	99.1	99.1	0.2
2	150 %	58.5	57.92	99.0		
3	150 %	58.5	58.10	99.3		

day and using different column. The system suitability results obtained in method precision and intermediate precision are compared and tabulated in Table-8. The data shows that the system suitability parameters from different day analysis are comparable. A comparison of method precision and intermediate precision results is given in Table-9. The results indicate that the method precision results range from 97.9 to 99.7 % with an RSD of 0.6 % and the intermediate precision results range from 98.7 to 99.3 % with an RSD of 0.24 %. The statistics of  $n = 12$  results ( $RSD_{12} = 0.45 %$ ) shows that the proposed method is precise.

TABLE-8  
COMPARISON OF SYSTEM SUITABILITY PARAMETERS

System suitability parameters	Method precision	Intermediate precision
Retention time (min)	1.740	1.728
USP tailing	1.0	1.0
USP plate count	21034	22350
RSD (%) of five standard injections	0.3	0.4

TABLE-9  
COMPARISON OF METHOD PRECISION AND INTERMEDIATE PRECISION RESULTS

Sample No.	Assay (%)	
	Method precision	Intermediate precision
1	99.4	99.3
2	97.9	99.0
3	99.2	99.2
4	99.7	99.2
5	99.1	98.8
6	99.1	98.7
Mean (n = 6)	99.1	99.0
RSD (%) (n = 6)	0.62	0.24
Mean (n = 12)	99.0	
RSD (%) (n = 12)	0.45	

**Range:** The range of the analytical method is the working analyte (mycophenolate mofetil) concentration between which the method is precise, accurate and linear. From the above validation data, the range of the method falls between 50 to 150 % of analyte concentration ( $125 \mu\text{g mL}^{-1}$ ) in the sample for which it has been demonstrated to have a suitable level of precision, accuracy and linearity.

**Limit of detection and limit of quantification:** The limit of detection was defined as the lowest concentration of the analyte that can be detected with S/N ratio 3:1. The limit of quantitation was defined as the lowest concentration of the analyte that can

be quantified accurately and precisely with S/N ratio 10:1. The limit of detection (LOD) and limit of quantification (LOQ) were estimated by using slope and standard deviation method. The obtained LOD and LOQ concentrations were further verified by injecting mycophenolate standard at these concentrations (Table-10). The obtained S/N ratio met the criteria for LOD and LOQ as per the ICH guidelines (Q2R).

TABLE-10  
RESULTS OF LOD AND LOQ

Parameter	$\mu\text{g mL}^{-1}$	S/N Ratio
LOD	1.2566	5
LOQ	4.1887	14

**Solution stability:** A study to establish stability of analytical solution (standard and sample) was conducted at 2 to 8 °C. Standard and sample solutions were prepared as per the proposed method and an aliquot was kept in refrigerator for 24 h. After 24 h, the two solutions were chromatographed and the area response was compared with that of a freshly prepared standard. For assessing the sample solution stability, difference between values of initial and 24 h assays were calculated. The data obtained from the study are presented in Table-11.

TABLE-11  
STABILITY OF ANALYTICAL SOLUTIONS

Sample No.	Sample solution stability		
	Initial result (% w/w)	Results after 24 h (% w/w)	Absolute difference
Sample-1	98.7	99.1	0.4
Sample-2	98.9	99.0	0.1
Standard solution stability			
Standard peak name	–	Similarity factor	–
Mycophenolate mofetil	–	1.00	–

**Robustness:** Robustness study was performed on the chromatographic parameters which are susceptible to change during the preparation of solution, handling and operation of the instrument. The parameters studied are mobile phase flow rate, column oven temperature and mobile phase pH. The mobile phase flow rate was studied at  $\pm 10 %$  from the optimized flow rate. The column oven temperature was studied at  $\pm 5 ^\circ\text{C}$ . The variation in mobile pH was studied at  $\pm 0.2$  units. The response factors observed during the robustness study are retention time (RT), USP resolution, USP Plate counts and relative standard deviation. The data shows that the change in mobile phase flow rate and mobile phase pH has considerable



TABLE-12  
RESULTS OF ROBUSTNESS/RUGGEDNESS EXPERIMENT

S. No.	Robustness parameter	Actual condition	Altered condition	Retention time of mycophenolate (min)	USP resolution	USP plate count	RSD (%)
1	Control	–	–	1.790	2.4	23710	0.20
2	Flow (mL min <sup>-1</sup> )	0.3	0.27 0.33	1.901 1.545	2.5 2.1	21084 22133	0.32 0.30
3	Column oven temperature (°C)	45	40 50	1.789 1.677	1.7 2.5	19039 20334	0.43 0.44
4	Mobile phase pH	4.5	4.3 4.7	1.821 1.656	2.2 2.3	19304 21203	0.40 0.20

effect on the retention behaviour of mycophenolate mofetil and USP resolution between USP related compound B and mycophenolate mofetil peak. Rest of the other parameters do not have much impact on the system suitability parameters. Table-12 shows the compilation of the robustness parameters and response observed. A graphical representation of retention behaviour and USP resolution *versus* robustness parameters is shown in Fig. 14.

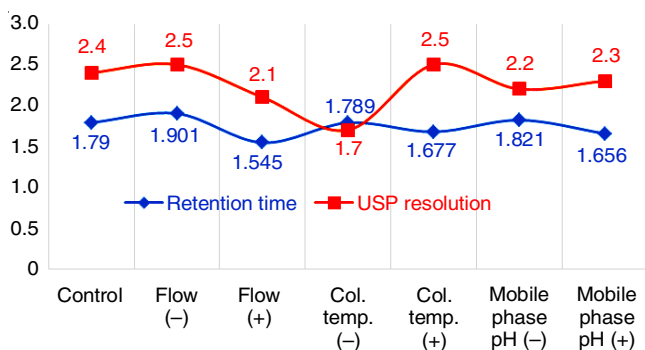


Fig. 14. Retention time and USP resolution vs. robustness parameters

## Conclusion

A rapid, simple, sensitive, accurate and reliable RP-UPLC method was developed and validated for the determination of mycophenolate in injection formulations as per the ICH guidelines. In this method mycophenolate mofetil was separated from its seven impurities in 9 min of chromatographic run time and quantified with high degree of accuracy and precision. The retention time for mycophenolate was found to be  $1.7 \pm 0.2$  min. Another advantage with this method is, it can be employed for mass determinations on LC-MS system, as its optimized buffer and mobile phase are compatible to mass spectrometry. The stability indicating power of the method was established through stress studies. All the degradation products formed during stress studies were well separated from the analyte peak which is evident from the peak purity data. The method discusses the nature of the molecule under stress conditions where it was shown that the compound is sensitive towards base hydrolysis (68.1 %), followed by oxidation (10 %), acid hydrolysis (3.7 %) and thermal degradations (2 %). The validation data meets the acceptance criteria for the validation parameters as per the current ICH (Q2R) and AOAC guidelines.

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