

# What is the magnetic moment of electron spin?

**Michael Tzoumpas**

Mechanical and Electrical Engineer  
National Technical University of Athens  
Irinis 2, 15234 Athens, Greece  
E-mail: m.tzoumpas@gmail.com

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**Abstract.** According to the unified theory<sup>1,2</sup> of dynamic space the inductive-inertial phenomenon<sup>3</sup> has been developed, forming the grouping<sup>3</sup> units (namely electric charges or forms of the electric field). Moreover, with the surface electric charges<sup>4</sup> of the electron cortex its inverse electric fields<sup>5</sup> are formed. By the above phenomena the actual theoretical value of the magnetic dipole moment of electron spin<sup>4</sup> is proved as equal to the experimental measurement.<sup>6</sup>

*Keywords:* Inductive phenomenon; grouping units; inverse electric field.

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## 1. Bohr's magneton

The magnetic dipole moment of the orbital electron

$$\mu_B = \frac{e\hbar}{2m}, \quad (1)$$

the so called Bohr's magneton,<sup>6</sup> has been calculated from

$$\mu = IA, \quad (2)$$

where

$$I = \frac{e}{T} = i \quad (3)$$

is the electric intensity,

$$A = \pi r^2 \quad (4)$$

the circle area,  $e$  the electric charge of the electron,  $r$  its orbital radius and  $T$  its orbital period (Fig. 1a). Therefore, due to Eq. 3, the Eq. 2 becomes

$$\mu = IA = iA = \frac{A}{T}e \Rightarrow \mu = \frac{A}{T}e. \quad (5)$$

The orbital momentum of the electron is

$$L = pr = mur \Rightarrow L = mur, \quad (6)$$

where  $p$  its impulse momentum,  $m$  its mass and

$$u = \frac{2\pi r}{T} \quad (7)$$

its peripheral speed. Therefore, due to Eqs 4 and 7, the Eq. 6 becomes

$$L = mur = m \frac{2\pi r}{T} r = 2m \frac{\pi r^2}{T} = 2m \frac{A}{T} \Rightarrow \frac{A}{T} = \frac{L}{2m} \quad (8)$$

and replacing in Eq. 5 it is

$$\mu = \frac{eL}{2m}. \quad (9)$$

However, the quantum orbital momentum is

$$L = \hbar = \frac{h}{2\pi}, \quad (10)$$

where  $h$  is the Planck's constant. Therefore, Eq. 9 becomes  $\mu_B = e\hbar/2m$  (Eq. 1), which is the quantum magnetic dipole moment of the orbital electron (Bohr's magneton).

## 2. Magnetic dipole moment of electron spin

For calculation<sup>6</sup> of the magnetic dipole moment of the electron spin, it has been considered that its electric charge  $e$  is the sum of elementary electric values  $q_i$  ( $e = \Sigma q_i$ ) with elementary mass  $m_i$  ( $m = \Sigma m_i$ ) and elementary spin momentum  $L_i$  ( $S = \Sigma L_i$ ). So, due to Eq. 9, it is

$$\mu_i = \frac{q_i L_i}{2m_i}. \quad (11)$$

Therefore,

$$\mu_s = \Sigma \mu_i = \frac{eS}{2m} \Rightarrow \mu_s = \frac{eS}{2m} \quad (12)$$

is the theoretical magnetic dipole moment of the electron spin, according to the modern Physics.<sup>6</sup>

However, the experimental measurement of the magnetic dipole moment of electron spin was found twice the above theoretical value and equal to

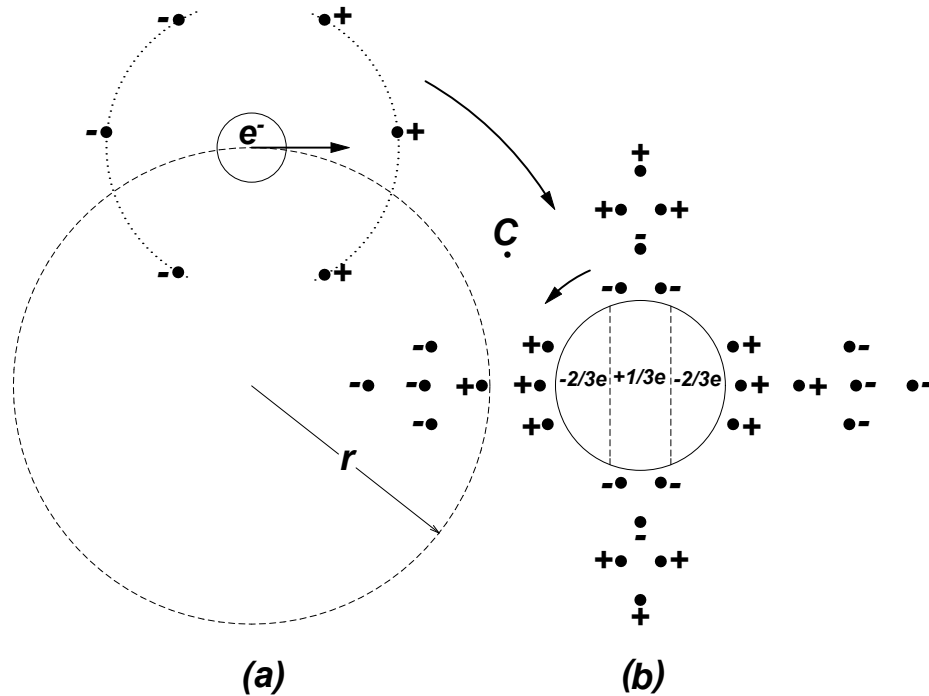
$$\mu_s = \frac{eS}{m}, \quad (13)$$

namely without the number 2 in the denominator, the so called g-factor.<sup>6</sup>

### 3. Theoretical proof the magnetic dipole moment of electron spin

The unified theory<sup>1,2</sup> of dynamic space proves the experimental value the magnetic dipole moment of the electron spin as follows:

The magnetic dipole moment of the electron spin<sup>4</sup> is interpreted with the surface electric charges of the electron cortex,<sup>7</sup> which are naturally present in its cortex as electric charges. These electric charges form inverse electric fields,<sup>5</sup> whose the polar diagrams are shown in Fig. 1b. So, the negative poles  $-2/3e$  of electron have near them the positive electric units<sup>8</sup> of the inverse electric field and beyond them the negative units of the external (common) field. The opposite occurs with the positive zone  $+1/3e$  of electron, whereby the electric fields are alternated. This alternate of positive and negative fields, which occurs during the electron spin and its passage from a place C, creates the magnetism<sup>3</sup> of the dipole moment.



**Figure 1.** Interpretation of the magnetic dipole moment of the electron spin,<sup>4</sup> with its grouping units<sup>3</sup> of the orbital electron (a) and with its inverse<sup>5</sup> electric fields of its surface electric charges<sup>7</sup> (b)

Moreover, by the above theory the experimental measurement ( $\mu_s = eS/m$ , Eq. 13) is interpreted as a result of the magnetism phenomenon created by the grouping units.<sup>3</sup> As the grouping units of the orbital electron rotate around the atomic nucleus (Fig. 1a) pass once from a place C to the time period T, when the opposite electric fields of the electron spin (Fig. 1b) pass twice from the same place C. Therefore, their passage time is  $T/2$ , namely they are rotated with a twice frequency, which implies a double magnetism for the electron spin.

However, the magnetic dipole moment of the orbital electron is equal to  $\mu = eA/T$  (Eq. 5). The  $e/T$  ratio corresponds to the electric intensity  $i$  of the orbital electron, which at the electron spin is twice the above intensity (due to the above passage time  $T/2$ ) and due to  $I = e/T = i$  (Eq. 3) we have

$$I = \frac{e}{T/2} = 2i. \quad (14)$$

Therefore, the magnetic dipole moment  $\mu = IA$  (Eq. 2) for electron spin, due to respective equation  $A/T = S/2m$  (Eq. 8), becomes

$$\mu_s = IA = 2iA = \frac{A}{T/2}e = \frac{2A}{T}e \Rightarrow \mu_s = \frac{eS}{m}. \quad (15)$$

So, we are found that the actual theoretical value (Eq. 15) of the magnetic dipole moment of electron spin, is equal to the experimental measurement (Eq. 13).

#### 4. The E/M wave has not magnetic dipole moment

The fundamental E/M wave<sup>9</sup> (Fig. 2) is constituted of two rotational spindle with electric charges<sup>10</sup>

$$Q = \pm e \quad (16)$$

for each spindle with a total spin

$$s = \pm 1, \quad (17)$$

of a constant photon length<sup>9,11</sup>

$$L = 3000m = 10^{58}L_0 \quad (18)$$

and a constant helix length<sup>9,11</sup>

$$\pi L = \pi 3000m = \pi L_0 10^{58}, \quad (19)$$

where

$$L_0 = 10^{-54}m \quad (20)$$

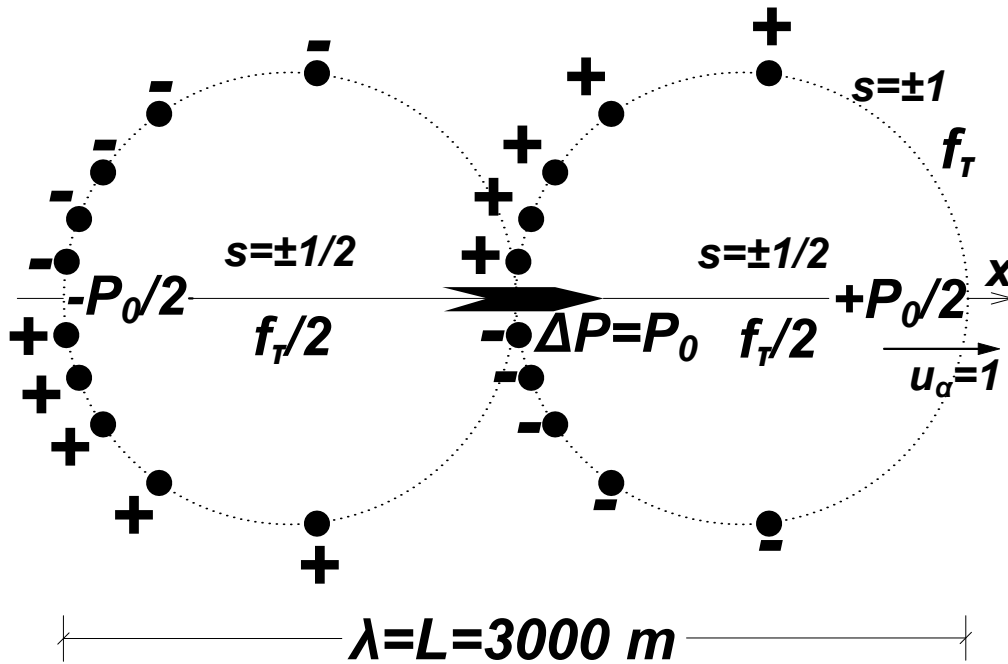
the length of the electric dipole.<sup>8</sup> In this helix length one or more forces talantonion<sup>12</sup>

$$f_\tau = 10^{26}N \quad (21)$$

can be accumulated that determine the number of the fundamental E/M waves and the wavelength  $\lambda$  of the photon.

Therefore, all fundamental E/M waves, which are derived from the motion meridians<sup>9,12</sup> of the electron and have a constant photon length  $L$  and a wave length  $\lambda$ , compose the autonomous motion of E/M waves,<sup>9</sup> the so called photon. These fundamental E/M waves (Fig. 2), which compose the photon, have a interchangeable spin ( $s = +1$  or  $s = -1$ ) and depending on their number (odd or even) the photon spin becomes  $s = \pm 1$  (Eq. 17) or

$$s = 0. \quad (22)$$



**Figure 2.** The two E/M formations compose the fundamental E/M wave<sup>9</sup> with electric charges<sup>10</sup>  $Q = \pm e$ , a spin  $s = +1/2+1/2 = +1$  or  $s = -1/2-1/2 = -1$ , an accumulated force of one talantonion<sup>9,12</sup> ( $f_\tau = f_\tau/2 + f_\tau/2$ ) and a wavelength<sup>9</sup>  $\lambda = L = 3000\text{m}$

So, the E/M wave is constituted of rotational spindles, whose the opposite electric charges ( $Q = \pm e$ , Eq. 16) give zero magnetic dipole moment. Therefore, the E/M wave is the only formation that has a spin and not magnetic dipole moment, but it has the extra spin value  $s = 0$  (Eq. 22).

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